

ACTIVATION OF MIXED CARBOXYLIC  $\alpha$ -BROMOTOLUOYL ANHYDRIDES BY SILVER  
 TETRAFLUOROBORATE. SYNTHESIS OF ESTERS AND THIOL ESTERS

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Summary: Mixed carboxylic  $\alpha$ -bromotoluoyl anhydrides were activated by silver tetrafluoroborate through intramolecular cyclization and reacted with alcohols or thiols giving the corresponding esters or thiol esters in good yields with the elimination of phthalide.

We previously reported a new type of mixed anhydrides (2) prepared from carboxylic acids and  $\alpha$ -bromotoluoyl bromide (1),<sup>1)</sup> which on heating at 80~145 °C in benzene or *N,N,N',N'*-tetramethylurea with alcohols gave carboxylic esters in good yields with the elimination of phthalide.<sup>2)</sup> In this communication are described a new, non-thermal activation of this unique mixed anhydrides and its application to the synthesis of esters<sup>3)</sup> and thiol esters.<sup>4)</sup>

In a search for effective additives which make the anhydrides usable under mild conditions as a general and strong acylating agents, a stoichiometric amount of silver tetrafluoroborate has been found to be the most satisfactory one. Activation of 2 occurred very rapidly in dichloromethane or acetonitrile at room temperature, and the subsequent addition of alcohols or thiols led to the formation of the corresponding esters or thiol esters in good yields. The results obtained are shown in Table 1. For the acylation of benzenethiol, however, the use of dimethoxyethane (DME) throughout the activation and esterification processes gave the best result. It is noteworthy that no epimerization was detected in the esterification of methyl hydrogen *meso*-2,4-dimethylglutarate. Cyclized cationic species (3) are postulated as the active intermediates for the reaction and the facile elimination of phthalide (4) from them is considered to promote their acyl-transfer property.

The experimental procedure is exemplified by the preparation of *S*-isopropyl 2-methylpentanethioate. The acylation of alcohols was carried out in the same manner.

$\alpha$ -Bromo-*o*-toluoyl bromide (1, 75  $\mu$ l, 0.5 mmol) was added to a stirred mixture of 2-methylpentanoic acid (63  $\mu$ l, 0.5 mmol), triethylamine (70  $\mu$ l, 0.5 mmol), and ether (0.8 ml). After 1 h, the mixture was filtered through Celite, concentrated, and diluted with acetonitrile (0.8 ml). To this solution were added silver tetrafluoroborate (92 mg, 0.5 mmol) and, after

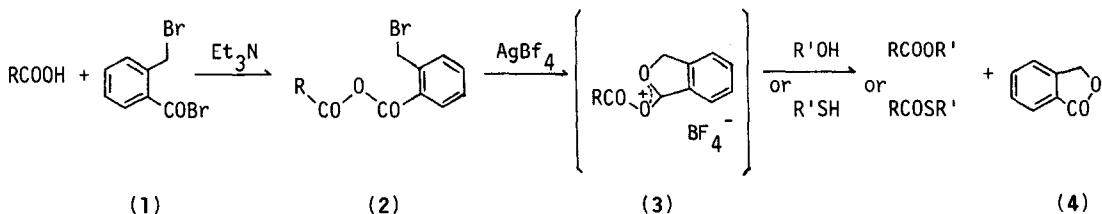


Table 1. Reaction of Mixed Anhydrides (2) Activated by Silver Tetrafluoroborate with Alcohols and Thiols

R in RCOOH	R'OH or R'SH'	Solvent	Time(min) <sup>a)</sup>	Yield <sup>b)</sup> of Ester or Thiol Ester
CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub>	(CH <sub>3</sub> ) <sub>2</sub> CHOH	CH <sub>2</sub> Cl <sub>2</sub>	8	84(67)
CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> (CH <sub>3</sub> )CH	CH <sub>3</sub> CH <sub>2</sub> OH	CH <sub>2</sub> Cl <sub>2</sub>	10	81(72)
CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> (CH <sub>3</sub> )CH	(CH <sub>3</sub> ) <sub>2</sub> CHOH	CH <sub>2</sub> Cl <sub>2</sub>	30	85(75)
(CH <sub>3</sub> ) <sub>3</sub> C	cyclo-C <sub>6</sub> H <sub>11</sub> OH	CH <sub>2</sub> Cl <sub>2</sub>	15	94
(CH <sub>3</sub> ) <sub>3</sub> C	C <sub>6</sub> H <sub>5</sub> OH	CH <sub>2</sub> Cl <sub>2</sub>	1	77(69)
C <sub>6</sub> H <sub>5</sub>	(CH <sub>3</sub> ) <sub>2</sub> CHOH	CH <sub>2</sub> Cl <sub>2</sub>	3	(62)
meso-CH <sub>3</sub> OCOCH(CH <sub>3</sub> )CH <sub>2</sub> (CH <sub>3</sub> )CH	(CH <sub>3</sub> ) <sub>2</sub> CHOH	CH <sub>2</sub> Cl <sub>2</sub>	40	84 <sup>c)</sup>
CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub>	(CH <sub>3</sub> ) <sub>3</sub> CSH	CH <sub>2</sub> Cl <sub>2</sub>	4	94(74)
CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> (CH <sub>3</sub> )CH	CH <sub>3</sub> CH <sub>2</sub> SH	CH <sub>3</sub> CN	5	(83)
CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> (CH <sub>3</sub> )CH	(CH <sub>3</sub> ) <sub>2</sub> CHSH	CH <sub>3</sub> CN	10	83(80)
(CH <sub>3</sub> ) <sub>3</sub> C	(CH <sub>3</sub> ) <sub>3</sub> CSH	CH <sub>3</sub> CN	15	(81)
(CH <sub>3</sub> ) <sub>3</sub> C	C <sub>6</sub> H <sub>5</sub> SH	DME	15	82 <sup>d)</sup>
meso-CH <sub>3</sub> OCOCH(CH <sub>3</sub> )CH <sub>2</sub> (CH <sub>3</sub> )CH	(CH <sub>3</sub> ) <sub>3</sub> CSH	CH <sub>2</sub> Cl <sub>2</sub>	5	72 <sup>c)</sup>

a) Reaction time required for alcoholysis or thiolysis of 2. b) Yields were determined by GLPC by comparing with appropriate internal standards. Isolated yields were given in parentheses. All the products were identified by <sup>1</sup>H-NMR by comparing with authentic specimens. c) No epimerization was detected by GLPC analysis. d) The reaction were carried out at -10 °C.

3 min, 2-propanethiol (46 μl, 0.5 mmol). The mixture was stirred for 10 min at room temperature, diluted with pentane (2 ml), and filtered. The filtrate was washed with saturated aqueous sodium hydrogencarbonate and water successively, dried over anhydrous sodium sulfate, and concentrated to a residue which was distilled to afford S-isopropyl 2-methylpentanethioate in 80% yield.

Further studies on the application of this type of anhydrides are in progress.

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